

Determination of the Protonation Constants and Study on Solvent Effect for Glycyl-asparagine in Different Aqueous Solutions of Methanol and at Constant Temperatures

Niloofer Soltani AHMADI¹, Farhoush KIANI^{2*}, Ahmad SHAHIDI¹ & Azadeh G. Hasan SARAEI¹

¹ Department of Food Science and Technology, College of Agriculture and Food Science, Ayatollah Amoli Branch, Islamic Azad University, Amol, Iran

² Department of Chemistry, Faculty of Science, Ayatollah Amoli Branch, Islamic Azad University, Amol, Iran

SUMMARY. In this work, the protonation constants for deprotonation processes of glycyl-asparagine, K_1 , K_2 , were determined in binary mixed solvents of water-methanol, containing 0, 10, 20, 30, 40, 50, 60, 70, and 80 % (v/v) methanol, at $T = 298.15$ K and constant ionic strength (0.1 mol.dm^{-3} NaCl). Determined data were analyzed using Kamlet, Abboud, and Taft parameters. In these processes, the spectrophotometric and potentiometric methods were used to determine the values of K_1 and K_2 .

RESUMEN. En este trabajo se determinaron las constantes de protonación para procesos de desprotonación de glycyl-asparagina, K_1 , K_2 , en solventes mixtos binarios de agua-metanol, conteniendo 0, 10, 20, 30, 40, 50, 60, 70 y 80 % (v/v) metanol, a $T = 298,15$ K y fuerza iónica constante ($0,1 \text{ mol.dm}^{-3}$ NaCl). Los datos determinados se analizaron utilizando los parámetros de Kamlet, Abboud y Taft. En estos procesos se utilizaron los métodos espectrofotométricos y potenciométricos para determinar los valores de K_1 y K_2 .

KEY WORDS: changes of enthalpy, changes of entropy, changes of Gibbs free energy, glycyl-asparagine, potentiometric, protonation constants, spectrophotometric.

* Author to whom correspondence should be addressed. *E-mail:* Farhoush_kiani@yahoo.com